

# Protic and Aprotic Ionic Liquids in Combination with Hard Carbon for Lithium-Ion and Sodium-Ion Batteries

Maria Arnaiz,<sup>[a]</sup> Peihua Huang,<sup>[b, c]</sup> Jon Ajuria,<sup>[a]</sup> Teófilo Rojo,<sup>[a, d]</sup> Eider Goikolea,<sup>\*,[d]</sup> and Andrea Balducci<sup>\*,[b, c]</sup>



- [a] M. Arnaiz, Dr. J. Ajuria, Prof. T. Rojo  
Power storage, Batteries and Supercaps  
CIC Energigune  
Albert Einstein 48, Technology Park of Araba, 01510 Miñao, Spain
- [b] Dr. P. Huang, Prof. A. Balducci  
Institute for Technical Chemistry and Environmental Chemistry  
Friedrich-Schiller-University Jena  
Philosophenweg 7a, 07743 Jena, Germany  
E-mail: andrea.balducci@uni-jena.de
- [c] Dr. P. Huang, Prof. A. Balducci  
Center for Energy and Environmental Chemistry Jena (CEEC Jena)  
Friedrich-Schiller-University Jena  
Philosophenweg 7a, 07743 Jena, Germany
- [d] Prof. T. Rojo, Dr. E. Goikolea  
Inorganic Chemistry Department  
University of the Basque Country (UPV/EHU)  
P.O. Box 644, 48080 Bilbao, Spain  
E-mail: eider.goikolea@ehu.es

In this work we report the use of the aprotic ionic liquid 1-butyl-1-methylpyrrolidinium bis(trifluoromethanesulfonyl)imide (Pyr<sub>14</sub>TFSI) and the protic ionic liquid 1-butylpyrrolidinium bis(trifluoromethanesulfonyl)imide (Pyr<sub>14H</sub>TFSI) in combination with olive pits derived hard carbon electrodes in view of the realization of advanced Li-ion and Na-ion batteries. We show that in the case of Li-based systems both classes of ionic liquids can be successfully utilized at room-temperature to build high performance and stable hard carbon-based systems. In the case of Na-ion chemistry, Pyr<sub>14</sub>TFSI appears very promising while the use of Pyr<sub>14H</sub>TFSI is not applicable due to the lack of electrochemical stability below 1 V vs. Na/Na<sup>+</sup>.